lon-chromatography (IC) is an established technique for analysis of inorganic and some organic ions. An IC column is an ion-exchange resin made of polar ionic groups chemically attached to the stationary support. A typical mobile phase for ion chromatography is water based with no or very little organic concentration. The stationary phase is completely wetted, and ion-exchange groups form solvated ions on a surface of the stationary support.

A significantly different interaction of the stationary phase with analytes is observed when ionexchange groups are shielded from direct interaction with aqueous mobile phase by a hydrophobic layer. This hydrophobic layer can be formed by brash type long alkyl chains attached to the same

We called the separation of ions on this covered hydrophobic ion-exchange stationary support the Shielded Ion-Exchange Liquid Chromatography (SIELC).

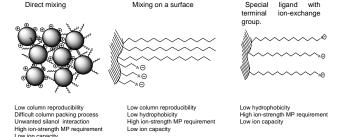
As opposed a traditional ion chromatography:

- · SIELC columns offer better selectivity and separation power due to a different type of ion interaction
- · SIELC usually requires less buffer concentration in the mobile phase to perform an ion-exchange process
- SIELC can operate both in low organic and high organic mobile phases:
- · SIELC selectivity and retention strongly depend on organic concentration in the mobile phase, which makes it possible to fine tune separations;
- · SIELC is compatible with a variety of detection techniques such as UV, IR, conductivity, electrochemical, MS and ELSD providing a great detection range and convenience of operation;
- SIELC demonstrates a higher ion loading capacity with a less concentrated buffer which is important for preparative applications and trace analysis;
- · Strong interaction of charged analytes with SIELC support allows retaining and analyzing polar charged compounds in reverse mode without ion-pairing reagents;
- · SIELC does not require any special IC equipment to analyze ions and can be performed using regular HPLC instrumentation.

Columns based on SIELC technology are, in fact, the reverse phase columns suitable for traditional RP separation. They also suitable for normal and HILIC separation.

Combining of two mechanism on one column

There are several ways of combining of two interactions on a single column



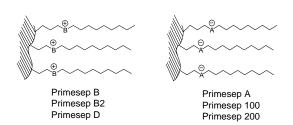
Chemical Structure of SIELC Based Columns

Special ligand with ion-exchange group in the middle of the alkyl chair

High column reproducibility

High column efficiency up to 100,000 plate/meter

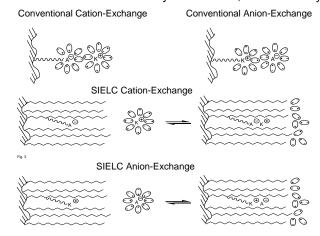
No silanol interaction, no high ion-strength MP requirement, high ion capacity, high hydrophobicity.



SIELC – New Approach to Ion Separations

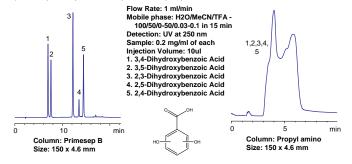
Yury Zelechonok, Vlad Orlovsky, SIELC





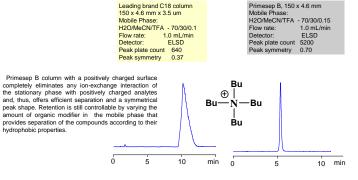
SIELC vs. IC Chromatography

When surface charge is not sufficiently solvated and interaction with ions occurs within non-polar environment it does not requires high ion-strength mobile phase to provide efficient chromatography. Primesep B and Propylamino columns have the same surface charge similar pKb of the ligand and similar density of the ligand on the silica surface, but they perform very differently with acidic analytes such as dihydrohybenzoic acids



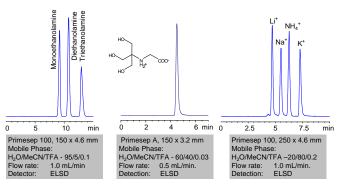
Column surface and analyte have the same charge

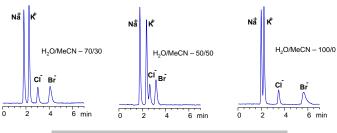
Leading brand C18 column



Separation of Ions in High Organic and Low Organic MP

SIELC technology allows to perform ion analysis in the mobile phase with high and low organic modifier concentration. This is important for analysis of complex mixtures where solubility of the sample can be a problem. Type and amount of organic modifier usually effect selectivity and retention.

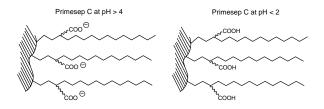




Serial connected Primesep 100, 50 x 4.6 mm & Primesep B, 50 x 4.6 mm Mobile Phase: H2O/MeCN with NH₄Ac 50 mM pH 5.0 Flow rate: 1.0 mL/min Detector: ELSD

SIELC Switch Phase[™] Technology

Columns based on SWITCH PhaseTM technology change their properties depending on pH of the mobile phase Embedded carboxylic acid is fully ionized at the pH above the transition point and loses charge when the pH of the mobile phase goes below the transition point. By controlling the pH of the mobile phase, the polar properties of the stationary phase can be altered to tune your separation needs.



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